Cooperative spin transition and thermally quenched high-spin state in new polymorph of [Fe(qsal)₂]I₃

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Abstract Novel polymorph of the ferric spin crossover complex, β -[Fe(qsal)₂]I₃ [qsalH = *N*-(8-quinolyl)salicylaldimine], has been prepared and characterized by magnetic susceptibilities and Mössbauer spectra. β polymorph exhibited a cooperative complete spin transition with a thermal hysteresis of 25 K and a tendency to be quenched in the high-spin (HS) state, which is contrastive to a gradual incomplete spin conversion in α polymorph.

Keywords Spin crossover · Ferric complex · Polymorph · Magnetic properties · Mössbauer spectra

1 Introduction

Polymorphism in spin crossover (SCO) compounds affords valuable information on the effects of the coordination geometries and intermolecular interactions upon their magnetic behaviors. Despite a number of examples of polymorphism in the Fe(II) SCO complexes, the polymorphic Fe(III) SCO complexes have been rarely reported [1, 2]. The family of the ferric SCO complexes, [Fe(qsal)₂]X [X = counter anion],

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was known as a versatile component for multifunctional SCO compounds [3–7]. Recently we have reported the crystal structures and gradual SCO behaviors under pressures for $[Fe(qsal)_2]I_3$ (Fig. 1) [8], which is hereafter designated as α - $[Fe(qsal)_2]I_3$. In the course of previous experiments, we have found another polymorph, β - $[Fe(qsal)_2]I_3$. In this paper we will report the temperature dependence of magnetic susceptibilities and Mössbauer spectra for β - $[Fe(qsal)_2]I_3$, which exhibited a cooperative spin transition with a wide thermal hysteresis of 25 K and a tendency to be quenched in the high-spin (HS) state.

2 Experimental

Preparation of β -[*Fe*(*qsal*)₂]*I*₃ Tetrabutylammonium triiodide (TBA·I₃) (80 mg) was dissolved in 160 ml of methanol and filtered off. The solution was poured into the filtered solution of [Fe(*qsal*)₂]NO₃ (50 mg) in 100 ml of methanol. The precipitates appeared within a few hours and were filtered off. β -[Fe(*qsal*)₂]I₃ was obtained as very thin dark brown platelets concomitant with a small amount of tiny black crystalline aggregates of α -[Fe(*qsal*)₂]I₃ and large black rectangular parallelepipeds, which was a nonstoichiometric compound designated as [Fe(*qsal*)₂]I_{2.75}. All the measurements were performed on the sample separated under microscope. Anal. Calcd for C₃₂H₂₂N₄O₂I₃: C, 41.28; H, 2.38; N, 6.02. Found: C, 41.01; H, 2.66; N, 5.80%.

Magnetic susceptibility measurements Magnetic susceptibilities were measured on a Quantum Design MPMS SQUID magnetometer under 0.5 T in the temperature range of 2–300 K. Data were corrected for diamagnetic contributions estimated by Pascal constants.

Mössbauer spectra The Mössbauer spectra were measured on a constant acceleration spectrometer with a source of ⁵⁷Co/Rh in the transmission mode. The measurements at low temperature were performed with a closed-cycle helium refrigerator (Iwatani Co. Ltd.). The obtained Mössbauer spectra were fitted with symmetric Lorentzian doublets by a least squares fitting program.

3 Results and discussion

3.1 Preparation and characterization

 β -[Fe(qsal)₂]I₃ was prepared by metathesis between [Fe(qsal)₂](NO₃) and TBA·I₃ in methanol. Contamination by α -[Fe(qsal)₂]I₃ and [Fe(qsal)₂]I_{2.75} sometimes occurred on more dilute or slower diffusion conditions. The composition for β -[Fe(qsal)₂]I₃ was confirmed by microanalyses. Powder XRD measurement for β -[Fe(qsal)₂]I₃ revealed that peak positions were different from the ones simulated from the crystal structure of α -[Fe(qsal)₂]I₃, clearly indicating that β -[Fe(qsal)₂]I₃ is a new polymorph of [Fe(qsal)₂]I₃.

3.2 Magnetic susceptibilities

The variable temperature magnetic susceptibilities were measured on a polycrystalline sample of β -[Fe(qsal)₂]I₃ at a scan speed of 1 K min⁻¹. The $\chi_M T$ versus T product is shown in Fig. 1. The $\chi_M T$ value at 300 K is 4.30 cm³ K mol⁻¹, suggesting the Fe(qsal)₂ cation is almost in the HS state. On lowering the temperature, the $\chi_{\rm M}T$ value abruptly decreased at around 95 K ($T_{1/2} \downarrow = 92$ K). The $\chi_{\rm M}T$ value of 0.60 cm³ K mol⁻¹ at 5 K indicated the complex is almost in the low-spin (LS) state. On heating the sample, the $\chi_M T$ value discontinuously increased at around 110 K ($T_{1/2} \uparrow = 117$ K). Since the temperature dependences of $\chi_M T$ on repeating the cooling and heating scans were reproducible, β -[Fe(qsal)₂]I₃ is a novel SCO compound showing a wide thermal hysteresis loop of 25 K. It should be noted that the $\chi_{\rm M}T$ values below the transition temperature depended on the cooling scan speeds. The faster the cooling speeds were, the higher the $\chi_M T$ values below the transition temperature were. When the room-temperature sample was inserted into an MPMS chamber setting the temperature of 5 K, the sample could be completely quenched in the HS state (Fig. 1, green triangles). On heating the quenched sample at a scan speed of 1 K min⁻¹, the quenched HS state was relaxed to the ground LS state at around 80 K. These magnetic behaviors were different from a gradual incomplete spin conversion without a thermal hysteresis in α -[Fe(qsal)₂]I₃ ($T_{1/2}$ = ca.240 K).

3.3 Mössbauer spectra

The Mössbauer spectra were measured on a polycrystalline sample of β -[Fe(qsal)₂]I₃. The spectrum at 293 K demonstrated one doublet (Fig. 2a), which was ascribed to the HS state (isomer shift (IS) = 0.232, quadrupole splitting (QS) = 0.616 mm s⁻¹). On cooling the sample without temperature control, the spectrum at 15 K consisted of one LS doublet (81.3%, IS = 0.062, QS = 2.861 mm s⁻¹) and one HS doublet (18.7%, IS = 0.331, QS = 0.595 mm s⁻¹) (Fig. 2b). Since the compound was easy to be trapped in the quenched HS state as described above, the spectrum at 15 K cooled after retaining the temperature of 80 K for a few hours was recorded again. Consequently, only one LS doublet spectrum was observed (IS = 0.075, QS = 2.878 mm s⁻¹) (Fig. 2c). This means that the former HS doublet is derived from the quenched HS state. On heating, the LS doublet (IS = 0.349, QS = 0.734 mm s⁻¹) was recovered at 130 K (Fig. 2d).





3.4 External treatments

When the crystals of β -[Fe(qsal)₂]I₃ were poured into pressure transmitting media such as Daphne 7373, Fluorinert, silicone oils, and water-ethanol, they broke into pieces and the spin transition disappeared. On the other hand, the magnetic behavior for α -[Fe(qsal)₂]I₃ have never changed in the pressure media [8]. Furthermore, grinding the crystals of β -[Fe(qsal)₂]I₃ also resulted in disappearance of the HS to LS conversion on lowering temperature. These observations indicate that the spin crossover behavior in β -[Fe(qsal)₂]I₃ is very sensitive to external treatments.

4 Conclusion

The temperature dependence of magnetic susceptibilities and Mössbauer spectra revealed that a new polymorphic ferric complex, β -[Fe(qsal)₂]I₃, exhibited a complete spin transition between the HS and LS states $(T_{1/2} \downarrow = 92 \text{ K}, T_{1/2} \uparrow = 117 \text{ K})$. This magnetic behavior was quite different from that of α -[Fe(qsal)₂]I₃ [8]. Furthermore, its magnetic behavior was quite sensitive to external treatments. These observations imply that a large structural change may accompany the spin transition in β -[Fe(qsal)₂]I₃. Since the comparison between the crystal structures of two polymorphs would provide the insight into a rational design for SCO compounds, the crystal structure analysis for β -[Fe(qsal)₂]I₃ is now in progress.

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References

- 1. Faulmann, C., Szilágyi, P., Jacob, K., Chahine, J., Valade, L.: New J. Chem. 33, 1268 (2009)
- Collet, E., Boillot, M.-L., Hebert, J., Moisan, N., Servol, M., Lorenc, M., Toupet, L., Cointe, M., Tissot, A., Sainton, J.: Acta Crystallogr. B65, 474 (2009)
- 3. Takahashi, K., Cui, H.-B., Kobayashi, H., Einaga, Y., Sato, O.: Chem. Lett. 34, 1240 (2005)
- Takahashi, K., Cui, H.-B., Okano, Y., Kobayashi, H., Einaga, Y., Sato, O.: Inorg. Chem. 45, 5739 (2006)
- Takahashi, K., Cui, H.-B., Okano, Y., Kobayashi, H., Mori, H., Tajima, H., Einaga, Y., Sato, O.: J. Am. Chem. Soc. **130**, 6688 (2008)
- 6. Takahashi, K., Mori, H., Kobayashi, H., Sato, O.: Polyhedron 28, 1776 (2009)
- Neves, A., Dias, J., Vieira, B., Santos, I., Branco, M., Pereira, L., Waerenborgh, J., Almeida, M., Belo, D., da Gama, V.: CrystEngComm 11, 2160 (2009)
- 8. Takahashi, K., Sato, T., Mori, H., Tajima, H., Sato, O.: Physica B 405, S65 (2010)