# An enthalpy of solution of rhenium in iron studied by <sup>57</sup>Fe Mössbauer spectroscopy

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**Abstract** The room temperature Mössbauer spectra of <sup>57</sup>Fe were measured for ironbased solid solutions  $Fe_{1-x}Re_x$ , with x in the range  $0.01 \le x \le 0.05$ . The obtained data were analysed in terms of the binding energy  $E_b$  between two rhenium atoms in the Fe-Re system. The extrapolated value of  $E_b$  for x = 0 was used for computation of enthalpy of solution of rhenium in iron. The result was compared with that resulting from the cellular atomic model of alloys by Miedema. The comparison shows that our findings are in a qualitative agreement with the Miedema's model predictions.

**Keywords** Mössbauer spectroscopy · Binding energy · Enthalpy of solution · Iron alloys

## **1** Introduction

The <sup>57</sup>Fe Mössbauer spectroscopy is a useful tool for the study of interactions of impurity atoms dissolved in iron [1–8]. At the same time from the findings concerning the impurity interactions one can easily derive the enthalpy of solution of the impurity elements in iron [9]. This considerably increases importance of the studies as the experimental values of the enthalpy play an essential role in developing and testing different models of binary alloys as well as methods for calculating the alloy parameters [10–16]. The main source of such experimental data is calorimetric study of the enthalpy of formation of binary systems [17]. Unfortunately, sometimes there are significant discrepancies in the data obtained by different authors. This is observed e.g. for the Fe-V alloys [18, 19]. Moreover, the calorimetric investigations are

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performed in relatively high temperatures at which some of iron systems are in their high-temperature  $\gamma$  (fcc) phases. Such situation exists for instance in the case of the Fe-Mn system. Consequently, there is no calorimetric data concerning the enthalpy of solution of Mn in the low-temperature  $\alpha$  (bcc) phase of Fe. All the above encouraged us to use the <sup>57</sup>Fe Mossbauer spectroscopy for supplying the experimental enthalpy of solution of different elements in  $\alpha$ -Fe. In our previous studies we collected the proper spectra for the Fe-V [1], Fe-Cr [2], Fe-Mn [3], Fe-Al [4], Fe-Ni [5], Fe-Co [5], and Fe-Mo [6] alloys.

In this work we have extended the investigations to the Fe-Re system. According to our knowledge there is no experimental data concerning the enthalpy of solution of Re in Fe. The value of the enthalpy can be simply predicted only on the basis of the semi-empirical cellular atomic model of alloys developed by Miedema [10].

x	$B_0$ [T]	$\Delta B_1[T]$	$\Delta B_2[T]$	$\Delta IS_1$ [mm/s]	$\Delta IS_2 \text{ [mm/s]}$
0.01	33.197(15)	-4.710(10)	-2.396(28)	-0.0348(10)	0.0104(10)
0.02	33.296(16)	-4.7524(63)	-2.502(13)	-0.0344(10)	0.0119(11)
0.03	33.396(14)	-4.8039(58)	-2.561(11)	-0.0320(10)	0.0154(11)
0.04	33.474(17)	-4.8152(63)	-2.609(10)	-0.0318(12)	0.0170(14)
0.05	33.523(19)	-4.8260(64)	-2.6490(98)	-0.0328(11)	0.0222(27)

**Table 1** Some of the best-fit parameters of the assumed model of the <sup>57</sup>Fe Mössbauer spectrum measured for  $Fe_{1-x}Re_x$  alloys

The standard uncertainties for the parameters result from the variance of the fit

#### 2 Experimental and results

The samples of iron-rhenium alloys containing 1, 2, 3, 4 and 5 at. % of Re, were prepared by melting appropriate amounts of the Aldrich 99.999% pure iron and 99.99% pure rhenium in an arc furnace filled with argon. Resulting ingots were cold-rolled to the final thickness of about 0.05 mm and then the foils were annealed in vacuum at 1270 K for 2 h. After that they were slowly cooled to room temperature during 6 h. Under these conditions, diffusion effectively stops at about 700 K [20], so the observed distributions of atoms in the annealed samples should be the frozen-in state corresponding to 700 K (the 'freezing' temperature  $T_d$ ). The room temperature measurements of the <sup>57</sup>Fe Mössbauer spectra for Fe-Re alloys were performed in transmission geometry by means of a constant-acceleration POLON spectrometer of standard design. The spectra measured are presented in Fig. 1.

They were analysed in terms of six six-line patterns corresponding to different hyperfine fields *B* at <sup>57</sup>Fe nuclei generated by different numbers of Fe and Re atoms located in the first two coordination shells of the probing nuclei. It was done under assumption that the influence of Re atoms on *B* as well as the corresponding isomer shift *IS* of a subspectrum, is additive and independent of the atom positions in the given coordination shells. In other words it was accepted that for each subspectrum the quantities *B* and *IS* are linear functions of the numbers  $n_1$  and  $n_2$  of Re atoms located, respectively, in the first and second coordination shells of <sup>57</sup>Fe and the functions can be written as follows:

$$B(n_1, n_2) = B_0 + n_1 \Delta B_1 + n_2 \Delta B_2,$$
  

$$IS(n_1, n_2) = IS_0 + n_1 \Delta IS_1 + n_2 \Delta IS_2,$$
(1)

where  $\Delta B_1$  ( $\Delta IS_1$ ) and  $\Delta B_2$  ( $\Delta IS_2$ ) stand for the changes of *B* (*IS*) with one Re atom in the first and second coordination shell of the Mössbauer probe. At the same time the quadruple splitting *QS* of a subspectrum is a free parameter [21]. It is worth noticing that the fits obtained under these assumptions are quite good and the found values of the best-fit parameters (displayed in Table 1) are in a good agreement with corresponding data given in the literature; e.g. in Ref. [22] one can find that for the Fe<sub>0.97</sub>Re<sub>0.03</sub> alloy  $\Delta B_1 = -4.72(3)$  T,  $\Delta B_2 = -2.25(7)$  T,  $\Delta IS_1 = -0.035(5)$  mm/s.

As the main result of the analysis the values of parameters  $c_1$  and  $c_2$  of each spectrum were determined. Assuming that the Lamb-Mössbauer factor is independent of the configuration of atoms in the surroundings of the <sup>57</sup>Fe nucleus, the parameters are total intensities of those components of a spectrum which are related to the

x	<i>c</i> <sub>1</sub>	<i>c</i> <sub>2</sub>	$p_1$	$p_2$	$E_{\rm b}~({\rm eV})$
0.01	0.1758(15)	-	0.1229	0.0081	_
0.02	0.2472(25)	0.0206(30)	0.2153	0.0286	0.033(10)
0.03	0.3084(27)	0.0464(33)	0.2827	0.0568	0.0219(55)
0.04	0.3329(28)	0.0898(33)	0.3294	0.0892	0.0003(31)
0.05	0.3417(25)	0.1246(31)	0.3593	0.1229	-0.0054

**Table 2** The binding energy  $E_b$  between a pair of Re atoms in  $Fe_{1-x}Re_x$  alloys deduced from the <sup>57</sup>Fe Mössbauer spectra

The standard uncertainties for  $c_1$  and  $c_2$  result from the variance of the fit of the assumed model to the spectrum measured. The values of uncertainty for  $E_b$  were computed assuming that the uncertainty for the 'freezing' temperature  $T_d$  is 50 K

Table 3 The enthalpy  $H_{\text{Fe-Re}}$  [eV/atom] of solution of rhenium in iron

Miedema's model [10]	This work $(\alpha$ -Fe)
-0.015	-0.22(13)

existence of one and two Re atoms in the two first coordination shells of nuclear probes <sup>57</sup>Fe, respectively.

$$c_1 = c (1, 0) + c (0, 1),$$
  

$$c_2 = c (2, 0) + c (1, 1) + c (0, 2).$$
(2)

The results are listed in Table 2.

The  $c_1$  and  $c_2$  values were used to calculate the binding energy  $E_b$  for pairs of Re atoms in the studied materials (Table 2). The computations were performed on the basis of the modified Hrynkiewicz-Królas formula [7, 23]

$$E_{\rm b} = -kT_{\rm d} \cdot \ln\left[(1 + 2 \cdot c_2/c_1) \cdot (c_2/c_1) \cdot (1 + 2 \cdot p(2)/p(1))^{-1} \cdot (p(2)/p(1))^{-1}\right].$$
(3)

In the next step we found the extrapolated value of  $E_b$  for x = 0 using  $E_b(0.02)$  and  $E_b(0.03)$ . Finally, the  $E_b(0)$  value was used for computation of the enthalpy of solution  $H_{\text{Fe-Re}}$  of Re in iron. The calculations were performed on the basis of the Królas model [9] for the binding energy according to which

$$H_{\rm Fe-Re} = -z \cdot E_{\rm b} \left(0\right) / 2 \tag{4}$$

where z is the coordination number of the crystalline lattice (z = 8 for  $\alpha$ -Fe). The result is displayed in Table 3 together with the value resulting from Miedema's model of alloys [10].

## **3** Conclusions

The positive value of the binding energy  $E_b$  between two rhenium atoms in the  $Fe_{1-x}Re_x$  alloys with  $x \le 0.03$ , speaks in favour of the suggestion that Re atoms interact repulsively in iron matrix. The value -0.22(13) eV/atom for the enthalpy of solution  $H_{Fe-Re}$  of rhenium in  $\alpha$ -iron, determined from the <sup>57</sup>Fe Mössbauer spectra are in a qualitative agreement with that -0.015 eV/atom, resulting from the semi-empirical Miedema's model of alloys [10] which is based on hundreds calorimetric data concerning the heat of formation for different binary systems. Both  $H_{\text{Fe}-\text{Re}}$  values mentioned above are negative but absolute values of them are essentially different.

Although till now, in the available literature, one cannot find proper theoretical calculations of  $H_{\text{Fe-Re}}$ , we hope that our experimental findings presented in this paper will stimulate such computations in future, just as it happened in the case of the Fe-Mo system [16].

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### References

- 1. Chojcan, J.: Interactions between V atoms in iron-based Fe-V solid solutions. J. Alloys Comp. **350**, 62–67 (2003)
- 2. Chojcan, J.: Interactions between Cr atoms in iron. Phys. Status Solidi B 219, 375-381 (2000)
- 3. Chojcan, J., Roztocka, G.: Interactions between Mn atoms in iron. Phys. Status Solidi B **204**, 829–833 (1997)
- Chojcan, J., Ostrasz, A.: A dilute-limit heat of solution of aluminium in iron studied with <sup>57</sup>Fe Mössbauer spectroscopy. Acta Phys. Pol. A **114**, 1573–1578 (2008)
- Idczak, R., Konieczny, R., Konieczna, Z., Chojcan, J.: An enthalpy of solution of cobalt and nickel in iron studiem with <sup>57</sup>Fe Mössbauer spectroscopy. Acta Phys. Pol. A 119, 37 (2011)
- Chojcan, J., Konieczny, R., Ostrasz, A., Idczak, R.: A dilute-limit heat of solution of molybdenum in iron studied with <sup>57</sup>Fe Mössbauer spectroscopy. Hyperfine Interact. 196, 377 (2010)
- 7. Chojcan, J.: Interaction between impurity atoms of 3d transition metals dissolved in iron. J. Alloys Compd. **264**, 50 (1998)
- Chojcan, J.: A dilute-limit heat of solution of 3d transition metals in iron studied with <sup>57</sup>Fe Mössbauer spectroscopy. Hyperfine Interact. 156/157, 523–529 (2004)
- Królas, K.: Correlation between impurity binding energies and heat of formation of alloys. Phys. Lett. A 85, 107–110 (1981)
- Miedema, A.R.: Energy effects and charge transfer in metal physics, modeling in real space. Physica B 182, 1–17 (1992)
- 11. Bangwei, Z., Yifang, O.: Theoretical calculation of thermodynamic data for bcc binary alloys with the embedded-atom method. Phys. Rev. B **48**, 3022–3029 (1993)
- Dursun, I.H., Güvenç, Z.B., Kasap, E.: A simple analytical EAM model for bcc metals. Commun. Nonlinear Sci. Numer. Simulat. 15, 1259–1266 (2010)
- Sluiter, M.H.F., Kawazoe Y.: Prediction of the mixing enthalpy of alloys. Europhys. Lett. 57, 526–532 (2002)
- 14. Klaver, T.P.C., Drautz, R., Finnis, M.W.: Magnetism and thermodynamics of defect-free Fe-Cr alloys. Phys. Rev. B 74, 094435 (2006)
- Bonny, G., Pasianot, R.C., Malerba, L., Caro, A., Olsson, P., Lavrentiev, M.Yu.: Numerical predictions of thermodynamic properties of iron-chromium alloys using semi-empirical cohesive models: the state of the art. J. Nucl. Mater. 385, 268–277 (2009)
- Lejaeghere, K., Cottenier, S., Claessens, S., Waroquier, M., Van Speybroeck, V.: Assessment of a low-cost protocol for an ab inito based prediction of the mixing enthalpy at elevated temperatures: the Fe-Mo system. Phys. Rev. B 83, 184201 (2011)
- Hultgren, R., Desai, P.D., Hawkins, D.T., Gleiser, M., Kelley, K.K.: Selected Values of Thermodynamic Properties of Binary Alloys. American Society for Metals, Metals Park (OH) (1973)
- 18. Myles, K.M., Aldred, A.T.: J. Phys. Chem. 68, 64 (1964)
- 19. Kubaschewski, O., Probst, H., Geiger, K.H.: Z. Phys. Chem. Neue Folge 104, 23 (1977)
- Cranshaw, T.E.: A Mössbauer study of <sup>119</sup>Sn in alloys of iron with Si, Al, and Rh: interaction potentials and phase diagrams. J. Phys. Condens. Matter 1, 829–846 (1989)
- Falepin, A., Cottenier, S., Comrie, C.M.: A. Vantomme, Interpreting Mössbauer spectra reflecting an infinite number of sites: an application to Fe<sub>1-x</sub>Si synthesized by pulsed laser annealing. Phys. Rev. B 74, 184108 (2006)
- Vincze, I., Campbell, I.A.: Mössbauer measurements in iron based alloys with transition metals. J. Phys. F 3, 647–663 (1973)
- Hrynkiewicz, A.Z., Królas, K.: Formation of two-impurity complex in dilute alloys observed through perturbed angular correlations of γ rays. Phys. Rev. B 28, 1864–1869 (1983)